STEREOCHEMISTRY OF THE Ph₃P-CC1₄ MEDIATED CYCLIZATION

OF CARBOXYLIC ACIDS AND 1,2-AMINO ALCOHOLS (VORBRUGGEN METHOD)

A. I. Meyers* and Denton Hoyer
Department of Chemistry, Colorado State University, Fort Collins, Colorado 80523

Summary: In contrast to a previous report, β -hydroxy amides formed via the Ph₃P-CCl₄ condensation of acids and amino alcohols, cyclize to oxazolines with complete inversion of the carbinol center.

The condensation of carboxyl groups and other nucleophiles with alcohols using $Ph_3P-DEAD$ has become an important technique in synthesis and is now known as the Mitsonobu reaction. Recently, this has been extended by Miller to the synthesis of β -lactams 2 by intramolecular cyclization of β -hydroxy hydroxamic acids 1 with $Ph_3P-DEAD$. The process proceeded with clean inversion at the carbinol center. Furthermore, the use of $Ph_3P-CCl_4-Et_3N$, previously utilized to transform alcohols to halides and amino alcohols to

1a, $R_1 = R_2 = H$ 1b, $R_1 = Me$, $R_2 = H$

1c, R₁ = H, R₂ = Me

aziridines, 4 also converted $\underline{1}$ to $\underline{2}$ with complete inversion. Another recent report by Vorbruggen 5 described a facile synthesis of oxazines, oxazolines, thiazolines, and imidazolines by employing Ph₃P-CCl₄-Et₃N to condense carboxylic acids with amino alcohols, mercaptoamines, and diamines, in a single step (Scheme 1). Although we have found this to be an excellent approach to these heterocycles, we were surprised to find the process differs from the mechanism claimed by Vorbruggen, but consistent with the claims of Miller. 2 We examined the oxazoline route described by Vorbruggen using 2-naphthoic acid and (2R, 3R)-3-amino-4-methoxy-2-butanol $\underline{7}$ obtained from threonine 6 and hoped to obtain the naphthyl oxazoline $\underline{9}$ needed for our studies on nucleophilic addition to chiral naphthalenes. 7 Instead, when the Vorbruggen conditions were applied, the only product

Scheme 1

obtained, in 75% yield, was the oxazoline 8 which had clearly undergone inversion at C-5. The cis alignment for C-4, C-5 in 8 was confirmed by an NOE determination. Irradiation of the C-5 methyl in 8 exhibited only a strong NOE for the vicinal proton at C-5 and no effect upon the distant proton at C-4. On the other hand, the trans-oxazoline 9 was prepared using the naphthylimidate, known to proceed with retention on 7 and the NOE showed the expected C-4 signal enhancement. Thus, we had confirmed the fact that the sequence described by Vorbruggen (Scheme I), indeed, proceeds with inversion when amino alcohols are employed using the $Ph_3P-CCl_4-Et_3N$ reagent. Apparently the Vorbruggen study was misled by the use of mercaptoamines 4 (X = S) or diamines 4 (X = NH) which produced heterocycles 5 containing sulfur and nitrogen. Since they did not use amino alcohols possessing a secondary alcohol, only primary alcohols, it could not be utilized as a probe for stereochemical results. Therefore, the intermediate hydroxy amide 3 (X = 0) reacts primarily with $Ph_3P^2-CCl_4^2$ at the hydroxyl group and displacement occurs by the amide carbonyl, as shown in 6, with inversion. The affinity of oxygen for phosphorous over that by nitrogen or sulfur is presumably the major reason that ring closure to 5 takes place in

CO₂H

$$(4R,5S) - \underline{8}$$

HO

$$H_2N$$

CH₂OMe

$$(2R,3R) - \underline{7}$$

Me

$$OEt$$

$$OH$$

$$(4R,5R) - \underline{9}$$

two different manners. As stated earlier, the work by Miller $(1 \longrightarrow 2)$ is consistent with the inversion of secondary alcohols under these conditions. Ironically, Miller also reported an oxazoline formation in his β -lactam work when $1 (Z = PhCH_2CO)$ was employed, but R_1 , R_2 were hydrogens (1a) and thus he was not in a position to detect the inversion

Scheme 2

during oxazoline ring closure (Scheme 2). Nevertheless, he suggested that since the only products isolated were β -lactam and oxazoline as shown in Scheme 2, that the oxyphosphonium salt is more likely to be the leaving group.

In summary, the Vorbruggen method to generate heterocycles is one of considerable utility, in view of the single step, one-pot reaction, but proceeds with high stereochemical control (inversion) when amino alcohols are employed and with retention when diamines or mercaptoamines are used.

References and Notes

- 1. O. Mitsunobu, Synthesis, 1 (1981).
- 2. M. J. Miller, P. G. Mattingly, M. A. Morrison, and J. F. Kerwin, Jr., <u>J. Am. Chem. Soc.</u> 102, 7026 (1980).
- 3. J. G. Calzada and J. Hooz, Org. Syn. 54, 63 (1974).
- 4. R. Appel and R. Kleinstuck, Chem. Ber. 107, 5 (1974); T. Okada, K. Schimura, and R. Sudo, <u>Bull. Chem. Soc. Japan</u> 43, 1185 (1970).
- 5. H. Vorbruggen and K. Krolikiewicz, Tetrahedron Letters 22, 4471 (1981).
- 6. Threonine was transformed via the oxazoline to the methoxyamino carbinol $\frac{7}{2}$ according to the following sequence. The overall yield of $\frac{7}{2}$ was 80%, mp 54-55°, $\left[\alpha\right]_{D}$ + 6.37° (c 5.7, CHCl₃).

- B. A. Barner and A. I. Meyers, J. Am. Chem. Soc. 106, 1865 (1984); A. I. Meyers and D. Hoyer, <u>Tetrahedron Letters</u> 25, 3667 (1984).
- 8. Vorbruggen (ref 5) attempted an oxazoline synthesis with a phenyl carbinol but reported no yield of product.

Acknowledgment: Financial support for this work was provided by the National Institutes of Health. AIM thanks the Alexander von Humboldt Foundation for a Senior Scientist Award (1984-1986) and the Faculty at the University of Wurzburg for their hospitality.

(Received in USA 24 June 1985)